

Atomic Structure Guided Notes Answers

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atoms of different elements have different masses. (all oxygen atoms have the same mass; and the mass of oxygen is always different from every other element's mass) • Compounds contain atoms of more than one element. (ex: NaCl) • In a particular compound, atoms of different elements always combine in the same way.

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- Write the atomic mass at the bottom of the square. Step 2: Determine the number of protons, neutrons, and electrons in each element. Step 3: Create an atomic structure diagram for each element. Step 4: Draw the Bonding Structure for each element. 5 B Boron 10.81 P = 5 N = 6 E = 5 Atomic Structure: Bonding Structure: B 5 P+ 6 no

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Democritus 4.

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Atomic Structure MODULE - 2 Notes Atomic Structure and Chemical Bonding chemistry has been defined as the study of matter in terms of its structure, composition and the properties. As you are aware, matter is made up of atoms, and therefore an understanding of the structure of atom is very important. You have studied in your earlier

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This product is a PowerPoint that covers the basics of the structure of an atom. It covers atoms as the building blocks of matter, protons, neutrons, electrons and their location within the atom, a brief explanation of electric charges, and the charges of the subatomic particles, and finally the concept of how the numbers of subatomic particles makes an atom unique from any other type of atom ...

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THOMSON'S ATOMIC MODEL Using what he had discovered, Thomson predicted what an atom should look like. These are the key points to Thomson's Atomic Model: 1. Because of its design this model is known as the plum pudding model 2. Each atom is a sphere filled with positively charged 'fluid'. This resembles the sticky jam part of a pudding. 3.

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The nonintegral atomic mass for naturally occurring chlorine is seen to be the weighted average of the atomic mass of its two major isotopes found by multiplying the atomic mass of each isotope times its decimal equivalent of its relative abundance: Now perform that calculation in the opposite direction.

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